

End Semester Examinations - 2015-16 Even Semester - May 2016

15PH3009 Atomic and Molecular Spectroscopy

Set A

Time : 3 hrs
Total Marks: 100

1.
 - a. Define spectroscopy. (3 marks)
 - b. Give the applications of different types of spectroscopy (7 marks)
 - c. Classify molecules base on moment of inertia (10 marks)
 - OR**
 2. Derive the rotational energy level (E_J) and rotational constant (B) of a rigid diatomic molecule with a neat sketch. (20 marks)
 3.
 - a. Give the difference between emission and absorption epectra (3 marks)
 - b. Discuss the vibrating diatomic molecue as a simple harmonic oscillator (17 marks)
 - OR**
 4.
 - a. What is meant by Molecular Polarizability (3 marks)
 - b. Explain Raman Spectra based on classical theory (17 marks)
 5. Explain the diatomic vibrating rotator (20 marks)
 - OR**
 6.
 - a. Discuss the fundamental vibrations of a CO_2 molecule (10 marks)
 - b. Discuss the fundamental vibrations of a H_2O molecule (10 marks)
 7.
 - a. Give an example of atom which has atomic spectra similar to hydrogen atom (2 marks)
 - b. With a neat energy level diagram explain the atomic spectra of hydrogen atom (18 marks)
 - OR**
 8.
 - a. What is J-J coupling (3 marks)
 - b. Explain Zeeman effect in detail with a neat sketch (17 marks)
 9. Explain the working principle, construction, and applications of XPS. (20 marks)
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Wishing you All the Best
